

PREPARATION OF SOME 2,3-DIHYDRO-7-OXO-7H-PYRIDO[1,2,3-*de*][1,4]BENZOXAZINE DERIVATIVES

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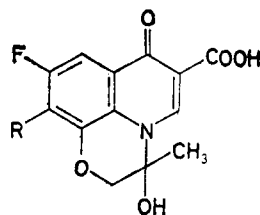
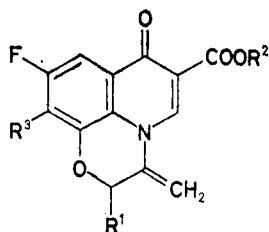
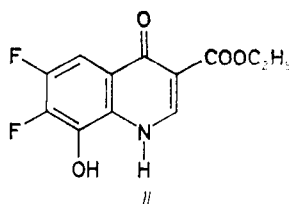
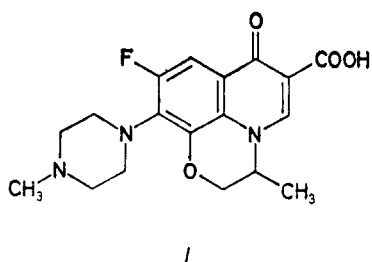
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In the last few years interest in analogs of ofloxacin (*I*) has considerably risen. We have recently described¹ a new reaction of 8-hydroxyquinolone *II* with 3-bromopropyne leading to 2,3-dihydro-3-methylene-7-oxo-7H-pyridol[1,2,3-*de*][1,4]benzoxazine-6-carboxylate *IIIa*, which could be used as an intermediate for preparation of antibacterically active compound *IIIb* (refs.^{1,2}).

Now we extend the scope of the reaction. Using 3-bromobutyne we obtained 2-methyl derivative *IIIc*. Treatment of *IIIc* with N-methylpiperazine followed by an



IIIa, $R^1 = H$, $R^2 = C_2H_5$, $R^3 = F$

IIIb, $R^1 = R^2 = H$, $R^3 = 4\text{-methylpiperazinyl}$

IIIc, $R^1 = CH_3$, $R^2 = C_2H_5$, $R^3 = F$

III d, $R^1 = CH_3$, $R^2 = H$, $R^3 = 4\text{-methylpiperazinyl}$

IVa, $R = F$

IV b, $R = 4\text{-methylpiperazinyl}$

alkaline hydrolysis provided ofloxacin analog *IIId*. On the other hand we have found that an acidic hydrolysis of *IIIA* provided *IVa* having a tertiary hydroxy group at position 3. When we treated *IIIA* with N-methyl-piperazine and the intermediate *IIIB* subjected the acidic hydrolysis we obtained 3-hydroxy analog of ofloxacin *IVb*.

EXPERIMENTAL

The melting points were determined on a Mettler FP 5 apparatus and were not corrected. IR spectra were taken on a Unicam SP-2006 spectrometer in KBr pellets; wavenumbers are given in cm^{-1} . UV spectra were taken on a Unicam 8 800 spectrophotometer in ethanol, molar absorption coefficients (ϵ) are given in $\text{m}^2\text{mol}^{-1}$, wavelengths (λ) in nm. Mass spectra were measured on MCH 1 320 and MAT 44 S spectrometers. ^1H NMR spectra (100 MHz) and ^{13}C NMR spectra (25.14 MHz) were measured on an apparatus BS-487 (Tesla Brno) 100 MHz in hexadeuterated dimethylsulfoxide (^{13}C NMR at 100°C , unless otherwise stated). The standard for ^1H NMR spectra was 3-trimethylsilylpropanoic acid, unless otherwise stated, the ^{13}C NMR spectra were referenced to tetramethylsilane. Chemical shifts are given in ppm (δ -scale), coupling constants (J) in Hz.

Ethyl 9,10-Difluoro-2,3-dihydro-3-methylene-2-methyl-7-oxo-7H-pyridol[1,2,3-de][1,4]benzoxazine-6-carboxylate (*IIIC*)

A solution of 3-bromobutyne (0.73 g, 5.5 mmol) was added dropwise during 1 h to a mixture of *II* (1.35 g, 5 mmol), sodium hydrogen carbonate (0.63 g, 7.5 mmol) and N,N-dimethylformamide (15 ml) at 90°C and then the mixture was stirred at this temperature for additional 2 h. The mixture was cooled down and poured onto ice (200 g). The insoluble portion was filtered off, washed with water and dried. Crystallization from 95% aqueous ethanol provided 0.9 g (56%) of *IIIB*, m.p. $171-173^\circ\text{C}$. For $\text{C}_{16}\text{H}_{13}\text{F}_2\text{NO}_4$ (321.3) calculated: 59.81% C, 4.08% H, 11.83% F, 4.36% N; found: 59.52% C, 4.10% H, 11.42% F, 4.01% N. ^{13}C NMR spectrum: 13.67 q (CH_3 of ethyl), 16.65 q (CH_3), 59.9 t (CH_2 of ethyl), 71.84 t (C-2), 102.41 t ($\text{CH}_2=\text{C}$), 103.16 d (C-8, $J_{\text{F,C}} = 20$), 112.75 s (C-6), 122.60 s* (C-7a), 124.50 s* (C-11a), 134.53 s (C-3), 135.60 s (C-11), 140.00 s (C-10), 141.20 d (C-5), 148.20 s (C-9), 163.52 s (COO), 171.35 s (C-7). UV spectrum, λ_{max} (log ϵ): 330 (3.25), 250 (3.05), 229 (3.41).

9-Fluoro-2,3-dihydro-2-methyl-3-methylene-10-(4-methyl-1-piperazinyl)-7-oxo-7H-pyrido[1,2,3-de][1,4]benzoxazine-6-carboxylic Acid (*IIId*)

A mixture of *IIIC* (0.6 g, 1.9 mmol), N-methylpiperazine (1 g, 10 mmol) and dimethyl sulfoxide (15 ml) was stirred at $110-120^\circ\text{C}$ for 13 h. The solution was evaporated in vacuo, the residue was suspended in ethanol (90 ml) and a solution of sodium hydroxide (0.4 g, 10 mmol) in water (10 ml) was added dropwise and the mixture was stirred at 50°C for 30 min. The formed solution was evaporated, the residue was dissolved in water (10 ml), acidified with hydrochloric acid, treated with charcoal and neutralized with 5% aqueous solution of sodium hydrogen carbonate. The mixture was extracted with trichloromethane, dried with magnesium sulfate and the residue after evaporation was crystallized from 95% aqueous ethanol: yields 0.4 g (50%), m.p. $271-273^\circ\text{C}$. For $\text{C}_{19}\text{H}_{20}\text{FN}_3\text{O}_4 \cdot 2.5 \text{H}_2\text{O}$ (418.4) calculated: 54.54% C, 6.02% H, 4.54% F, 10.04% N; found: 54.56% C, 5.24% H, 4.58% F, 9.96% N. IR spectrum: 2 407 (OH), 1 719 (COOH), 1 660 ($\text{CH}_2=\text{C}$), 1 617 ($\text{C}=\text{O}$). UV spectrum, λ_{max} (log ϵ): 336 (3.07), 299 (3.40), 244 (3.04), 220 (3.25).

^1H NMR spectrum (90°C): 1.65 d, 3 H (CH_3); 2.48 bt, 4 H ($\text{H}-3'$, $\text{H}-5'$, of piperazine); 2.84 s, 3 H (CH_3 of piperazine); 3.32 bt, 4 H ($\text{H}-2'$, $\text{H}-6'$ of piperazine); 5.14 m, 1 H (C-2); 5.54 d, 5.98 d, 2 H ($\text{CH}_2=\text{CH}$); 7.62 d, 1 H ($\text{H}-8$, $J_{\text{H,F}} = 11$); 8.90 s, 1 H ($\text{H}-5$). Mass spectrum, $m/z = 373$.

9,10-Difluoro-2,3-dihydro-3-hydroxy-3-methyl-7-oxo-7H-pyridol[1,2,3-de][1,4]benzoxazine-6-carboxylic Acid (*IVa*)

A suspension of *IIIa* (0.25 g, 0.8 mmol) in a mixture of concentrated hydrochloric acid (2 ml) and acetic acid (2 ml) was refluxed for 4 h. The mixture was cooled down and the separated crystals were filtered off and crystallized from N,N-dimethylformamide; yield 0.15 g (62%), m.p. $253-257^\circ\text{C}$. For $\text{C}_{13}\text{H}_9\text{F}_2\text{NO}_5$ (297.2) calculated 52.53% C, 3.05% H, 12.78% F, 4.71% N; found: 52.20% C, 3.15% H, 12.82% F, 4.49% N. IR spectrum: 3180 (OH), 1680 (COOH), 1570, 1520 (aromat. system). UV spectrum, λ_{max} (log ϵ): 317 (3.08), 232 (3.37). ^1H NMR spectrum (110°C): 1.76 s, 3 H (CH_3); 4.48 s, 2 H (C-2); 7.83 dd, 1 H ($\text{H}-8$, $J_{\text{H,F}} = 10.00$, 10); 9.10 s, 1 H ($\text{H}-5$). Mass spectrum, $m/z = 297$ (M^+).

9-Fluoro-2,3-dihydro-3-hydroxy-3-methyl-10-(4-methyl-1-piperazinyl)-7-oxo-7H-pyrido[1,2,3-de][1,4]benzoxazine-6-carboxylic Acid (*IVb*)

A mixture of *IIIa* (1 g, 3.25 mmol), N-methylpiperazine (1 g, 10 mmol) and dimethyl sulfoxide (20 ml) was stirred at $110-120^\circ\text{C}$ for 10 h. The solution was evaporated in vacuo, the residue was suspended in a mixture of 10% hydrochloric acid (10 ml) and acetic acid (10 ml) and the mixture was refluxed for 4 h. The mixture was evaporated to dryness, the residue was triturated with methanol (5 ml) and left to stand overnight in a refrigerator, insoluble portion was filtered off and crystallized from methanol; yield 0.4 g (28%), m.p. $251-253^\circ\text{C}$. For $\text{C}_{18}\text{H}_{20}\text{FN}_3\text{O}_5 \cdot \text{HCl} \cdot \text{H}_2\text{O}$ (431.8) calculated: 50.06% C, 5.37% H, 8.21% Cl, 4.40% F, 9.73% N; found: 50.02% C, 5.03% H, 8.45% Cl, 4.79% F, 9.88% N. IR spectrum: 2679 (OH), 1682 (COOH), 1618 (C=O). UV spectrum, λ_{max} (log ϵ): 297 (3.52). ^{13}C NMR spectrum: 24.43 q (CH_3), 41.98 q (N- CH_3), 46.46 t, 46.61 t (C-2', C-6' of piperazine), 52.81 t (C-3', C-5' of piperazine), 70.81 t (C-2), 83.62 s (C-3), 103.19 d (C-8, $J_{\text{F,C}} = 24$), 106.67 s (C-6), 119.85 s (C-7a), 124.68 s (C-11a), 129.79 s (C-10), 139.90 s (C-11), 140.05 d (C-5), 154.57 s (C-9, $J_{\text{H,F}} = 247$), 165.23 s (COO), 175.94 s (C-7). Mass spectrum, $m/z = 377$.

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